

Lithium-ion batteries, despite their high energy density, exhibit a gradual loss of charge even when not in use. This phenomenon, known as self-discharge, significantly impacts battery lifespan and performance. Understanding the underlying mechanisms of self-discharge is crucial for optimizing battery design and maximizing operational life.

Lithium battery self-discharge refers to the natural reduction in a battery's charge over time while in an open-circuit state (i.e., not connected to a load or charger). This charge loss is caused by internal micro-short circuits and unwanted chemical side reactions.

However, even when not in use, lithium batteries gradually lose their charge--a phenomenon known as self-discharge. While this is a natural characteristic of batteries, if left unmanaged, it can lead to performance degradation and even safety risks (explore li ion battery safety).

This phenomenon, known as self-discharge, significantly impacts battery lifespan and performance. Understanding the underlying mechanisms of self-discharge is crucial for optimizing battery design and maximizing operational life. Self-discharge refers to the spontaneous loss of battery capacity while in an open-circuit state.

Factors causing micro-shorts include: Poor sealing: Poor gasket or packaging sealing performance can lead to external electron leakage. Irreversible chemical side reactions directly deplete active lithium and electrolyte, resulting in power loss and shortened battery life. Mainly include:

External Electronic Leakage: Poor insulation of the battery seal,gasket,or external leads can lead to current leakage,contributing to self-discharge. Moisture: Moisture within the battery can react with the electrolyte (often containing lithium hexafluorophosphate (LiPF₆)) to generate hydrofluoric acid (HF).

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